
Fact Sheet

Uranium Hexafluoride (UF₆)

Table of Contents

1	Introduction	3
2	Preparation	3
3	Properties	4
4	Application	5
5	Transportation of UF ₆	6

01

Introduction

Uranium hexafluoride (UF₆) is a chemical compound used in the 'front end' of the nuclear fuel cycle. The 'front end' consists of the following steps; mining and milling, refining, conversion, enrichment and fuel fabrication. UF₆ is used in the steps conversion, enrichment and fuel fabrication. UF₆ is converted from uranium ore concentrate (UOC) (see WNTI Fact sheet) in conversion plants and is used in uranium enrichment facilities to produce enriched UF₆. The enriched UF₆ is converted to uranium oxide in fuel fabrication plants to produce fuel for nuclear reactors for electricity generation.

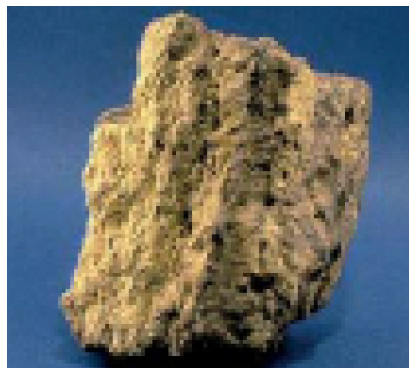


Image 1. - Natural Uranium Ore

02

Preparation

In a conversion plant the uranium ore concentrate (U₃O₈) is chemically converted to uranium tetrafluoride (UF₄). The UF₄ is then reacted with elemental fluorine (F₂) at elevated temperatures to create uranium hexafluoride (UF₆).

Because the commercial enrichment facilities use a physical process, the UF₆ does not undergo chemical changes during enrichment operations. The enriched UF₆ is chemically converted to uranium oxide (UO₂) in a fuel fabrication plant.



Image 2. - Uranium Hexafluoride

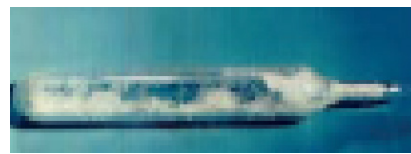


Image 3. - Uranium Hexafluoride

03

Properties

At 20°C, UF₆ is a solid, colourless, crystalline mass with a density of 5.06 g/cm³. At a temperature of 56.5°C, the vapour pressure is atmospheric (1 bar or 100kPa). The triple point of UF₆ is at 64 °C and 1.5 bar. At and above the triple point, UF₆ exists in the solid, liquid and gaseous phases. Below its triple point, UF₆ can only be a solid or a gas. The different phases and phase changes are shown in the diagram below.

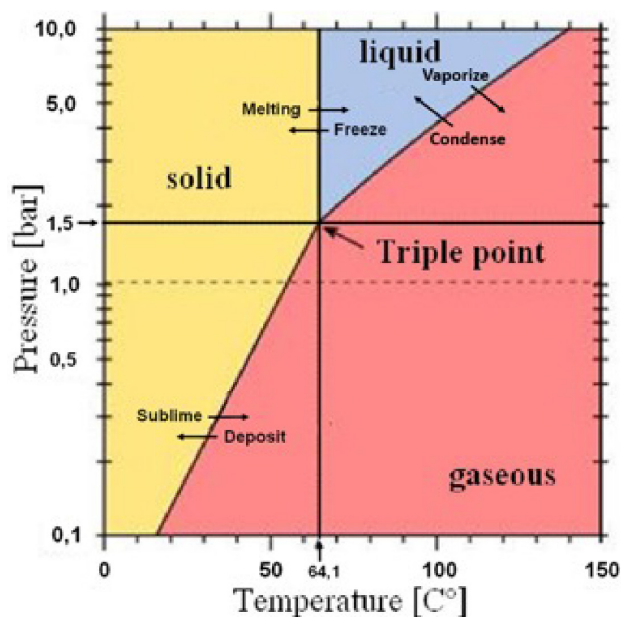


Image 4 - UF₆ Phase Diagram

UF₆ is a radioactive material. Its radioactivity derives from the uranium component. Uranium is mainly an alpha “α” and gamma “γ” emitter due to its characteristic decay products. There is a negligible beta “β” and neutron emission. A person standing for 2 hours next to a cylinder filled with natural UF₆ would receive approximately a 7 microsievert (μSv) effective dose, which is comparable to one dental X-ray or a 2.5-hour commercial flight.

Although UF₆ is classified as a low specific activity material (LSA) for transportation, it bears a subsidiary hazard because it produces corrosive and chemotoxic by-products on contact with moist air if released to the atmosphere. The reaction of UF₆ with water results in soluble uranyl fluoride (UO₂F₂) and gaseous hydrogen fluoride (HF) according to the following equation.



The gaseous HF with water (body fluid) forms hydrofluoric acid, which is a strong corrosive. The reaction products of UF₆ also have significant toxic properties.

For larger quantities > 100 g per package:

The main hazard for solid UF₆ in larger quantities, as determined by the transport regulations, is radioactivity followed by toxicity and corrosivity as subsidiary hazards. Since solid UF₆ sublimates when exposed to ambient air, the real predominant hazard to be considered shall be that of its gaseous form considering the large quantities of UF₆: corrosivity followed by toxicity and radioactivity.

For quantities < 100 g per package:

The main hazard for solid UF₆ sample quantities, as determined by the transport regulations, is toxicity followed by corrosivity and radioactivity as subsidiary hazards. Since solid UF₆ sublimates when exposed to ambient air, the real predominant hazard to be considered shall be that of its gaseous form considering the limited quantity of UF₆: toxicity followed by corrosivity and radioactivity.

04

Application

Natural uranium consists mainly of two isotopes ²³⁸U (99.3%) and ²³⁵U (0.7%). The fuel required by most nuclear reactors needs a ²³⁵U fissile isotope content of between 3.5 – 5%. This higher ²³⁵U content is achieved through a process known as “enrichment” by e.g. gas

centrifuges. In the past, the enrichment of UF₆ was also performed in diffusion plants. Due to extensively higher energy consumption, this technology is not cost effective and is not practised anymore.

UF₆ is the ideal chemical compound for the enrichment process because it exists in gaseous form at relatively low temperatures and pressures. In enrichment plants, centrifuges separate the fissile isotopes ²³⁵U and ²³⁸U in a physical process by using the slightly different masses of the isotopes. The result is a depleted UF₆ stream (²³⁵U 0.2-0.3%) and an enriched UF₆ stream (²³⁵U 3.5 – 5%).

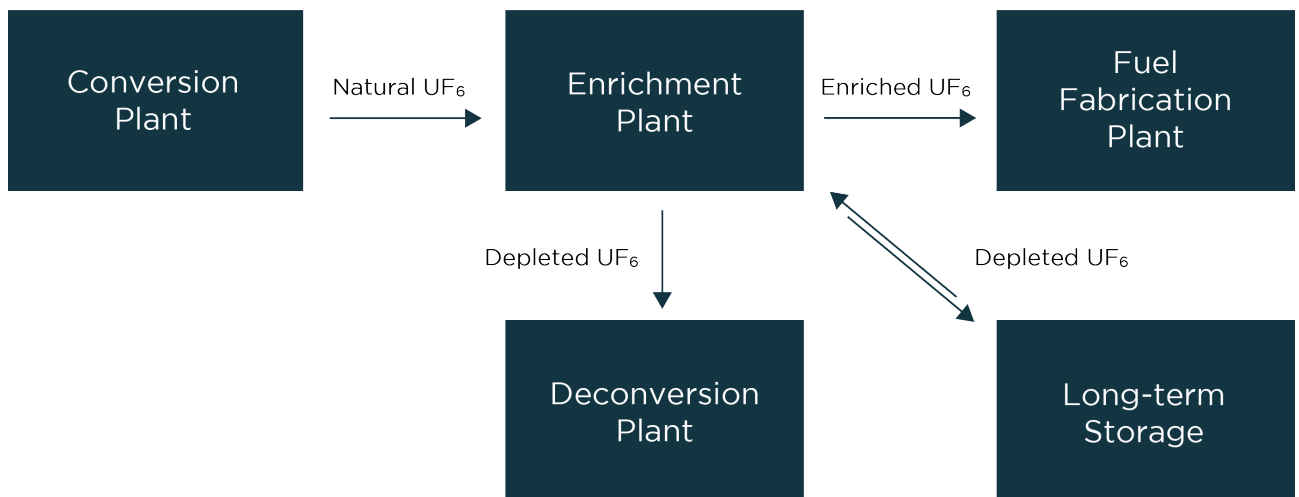


Image 5. - Transport of UF₆ between different industry branches

05

Transportation of UF₆

UF₆ is transported via road, rail, sea and air. International and national dangerous goods regulations apply to the transport of UF₆. For quantities of UF₆ greater than 0.1 kg the primary hazard is from the radioactivity and therefore classified as Class 7, with Class 6.1 (toxic) and Class 8 (corrosive) as subsidiary risks. The cylinders need to withstand a fire test for 30 minutes under well-defined conditions and fulfil definite leakage rates and resist drop



Image 6 - Type 30B Cylinder

30B Cylinder

Diameter: ~ 0,75m (30 inch)

Length: ~ 2m

Wall thickness: 12,7mm (1/2 inch)

Volume: ~ 750L

Tare weight: ~ 635kg

Max. fill: 2,277kg

Gross Weight: ~ 3,000kg

Proof pressure: 28 bar (5-year inspection)

Max. Operating temp: 121°C/250°F

tests. These requirements are defined in the “IAEA Regulations for the Safe Transport of Radioactive Material” and in the “United Nations Recommendations for the Transport of Dangerous Goods” as well as in the derived dangerous goods transport regulations around the world.

Transport Cylinders

Natural and depleted UF₆ is transported in 48Y cylinders, whereas for enriched UF₆, smaller 30B cylinders are used. Outer protection such as thermal protection or protective shipping packages (PSP) may be required to comply with the regulatory transport requirements. For shipments of sample quantities of UF₆, other packages are used.



Image 7 - Type 48Y Cylinder

48Y Cylinder

Diameter: ~ 1,25m (48 inch)

Length: ~ 4m

Wall thickness: 16mm (5/8 inch)

Volume: ~ 4.000L

Tare weight: ~ 2,500kg

Max. fill: 12,500kg

Gross Weight: ~ 15,000kg

Proof pressure: 28 bar (5-year inspection)

Max. Operating temp: 121°C/250°F

UF₆ is normally transported using standard multimodal container-based flatracks. These can be easily transferred between road, rail and maritime shipments. One flatrack can accommodate four 30B cylinders

loaded in protective shipping packages or one 48Y cylinder. 48Y cylinders are also transported as single units on road trailers, railway wagons and in ship holds.



Image 8 - 30B Cylinder with PSP



Image 9 - 48Y Cylinder with outer protection

Labelling for Transport

There are two UN numbers available for the transport of UF₆ for quantities greater than 0.1 kg, UN2977 is used for “Radioactive Material, Uranium Hexafluoride, fissile” and UN2978 is used for “Radioactive Material, Uranium Hexafluoride, non fissile or fissile excepted”. The third UN number UN3507 “Uranium Hexafluoride,

Radioactive Material, excepted package, less than 0.1 kg per package, non-fissile or fissile excepted” is only applicable to UF₆ quantities ≤ 0.1 kg.

Note - The transport regulations are cited according to international regulations and any national deviations are not considered in this Fact Sheet.

UN2977 RADIOACTIVE MATERIAL, URANIUM HEXAFLUORIDE, FISSILE



UN2978 RADIOACTIVE MATERIAL, URANIUM HEXAFLUORIDE, non fissile or fissile-excepted



UN3507 URANIUM HEXAFLUORIDE, RADIOACTIVE MATERIAL, EXCEPTED PACKAGE, less than 0.1 kg per package, non-fissile or fissile excepted



ASTM

ASTM International is an international standards organization that develops technical standards for a wide range of materials, products, systems, and

services. There are two ASTM specifications defined for UF₆. For commercial natural UF₆ ASTM C787 defines impurity and uranium isotope limits and additional limits for reprocessed UF₆. ASTM C996 is the comparable standard for enriched UF₆ up to 5% ²³⁵U.

Explanatory note:

Reprocessed uranium hexafluoride is not considered in this fact sheet.

References:

ASTM C787-15; ASTM C996-15; IAEA SSR-6; WNTI; world-nuclear.org; ead.anl.gov; rsc.org; nrc.gov



Whilst the WNTI will use all reasonable efforts to ensure that the information in this Fact Sheet is accurate, we cannot guarantee the accuracy of all information and we will accept no liability for any loss or damages incurred, howsoever caused, and cannot be held liable for any use or reliance you may make of or put on it. The WNTI also cannot be held liable for your use or inability to use the site or the information or services that it contains. Errors and Omissions Accepted.

The WNTI offers the use of this Fact Sheet freely to members and non-members of the transport community. Where any interpretation of the information has been made, it has been done so with the interests of the wider transport community. Although the standard has been extensively reviewed by industry experts, if you have any issues in use or content, please contact the WNTI so we can rectify the issues and conflicts in systems etc.

Aviation House
125 Kingsway
London, WC2B 6NH
United Kingdom

Tel: +44 (0)20 7580 1144
Fax: +44 (0)20 7580 5365

Web: www.wnti.co.uk
Email: wnti@wnti.co.uk

WNTI Fact Sheet Uranium Hexafluoride (UF₆)

© World Nuclear Transport Institute Registered in
England and Wales, Company Number 3557369

